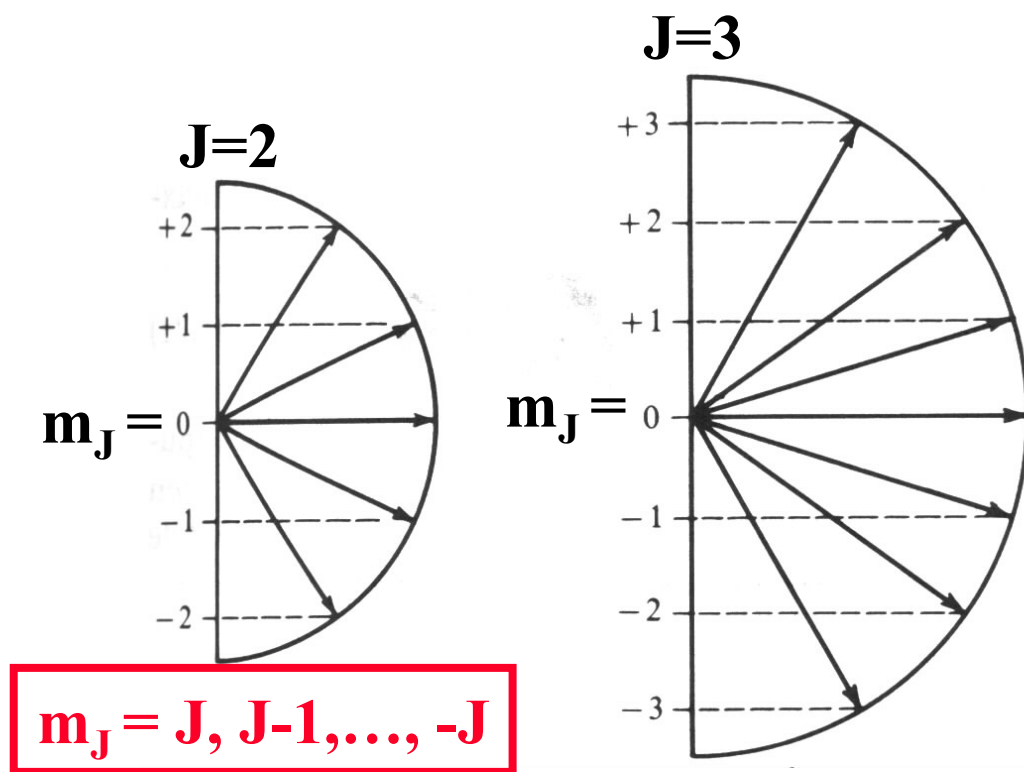


Degeneracy



Energy only
determined by J



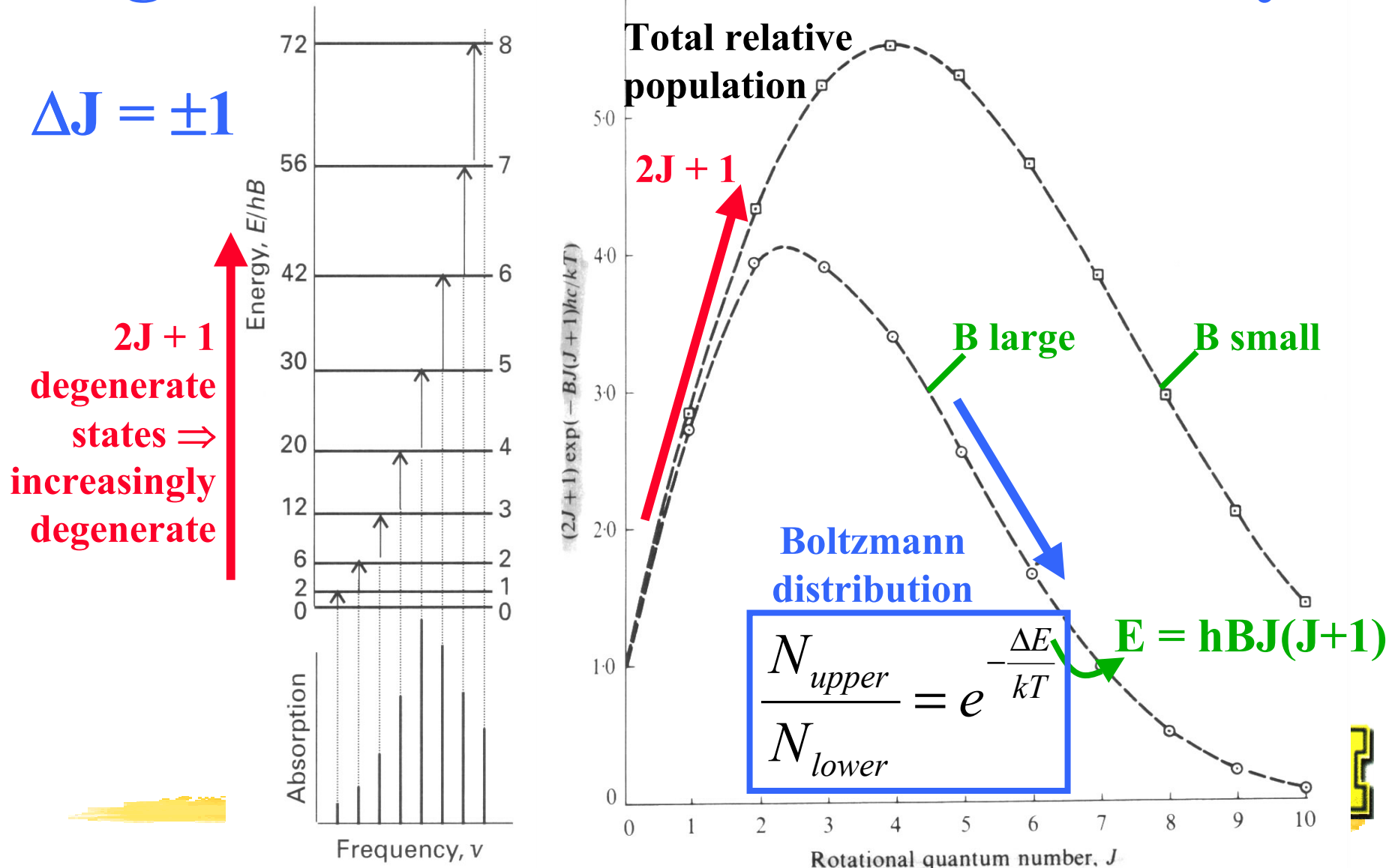
all $m_J = -J, \dots, +J$
share the same
energy



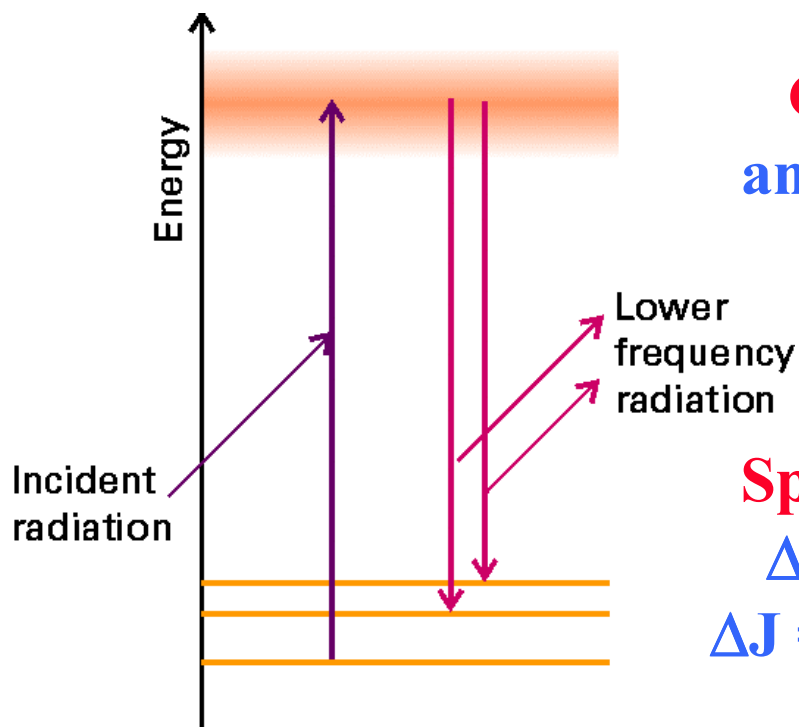
$2J+1$ degeneracy

Selection rule: $\Delta m_J = 0, \pm 1$

The allowed rotational transitions of a rigid linear rotor and their intensity

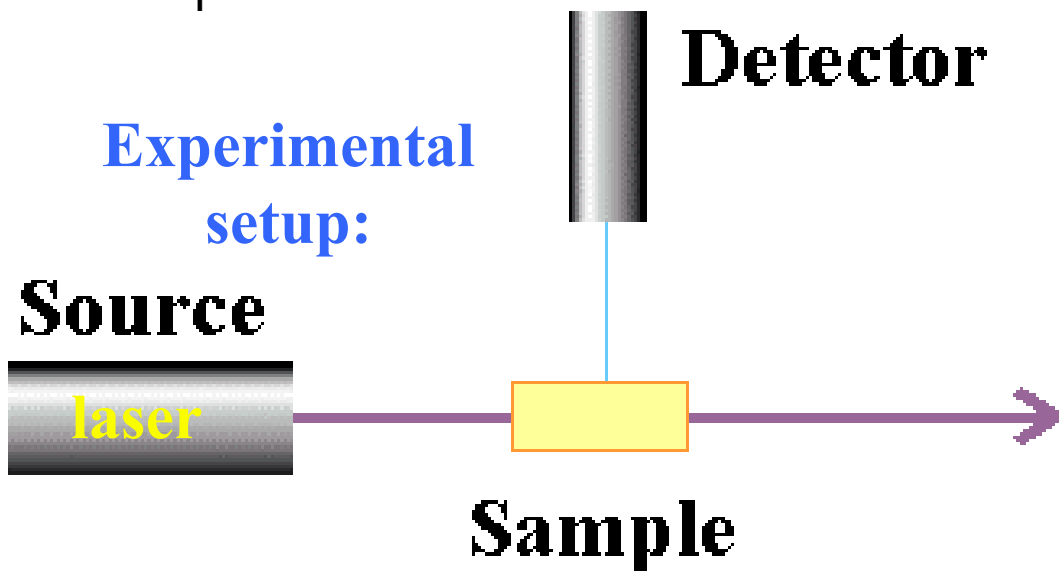
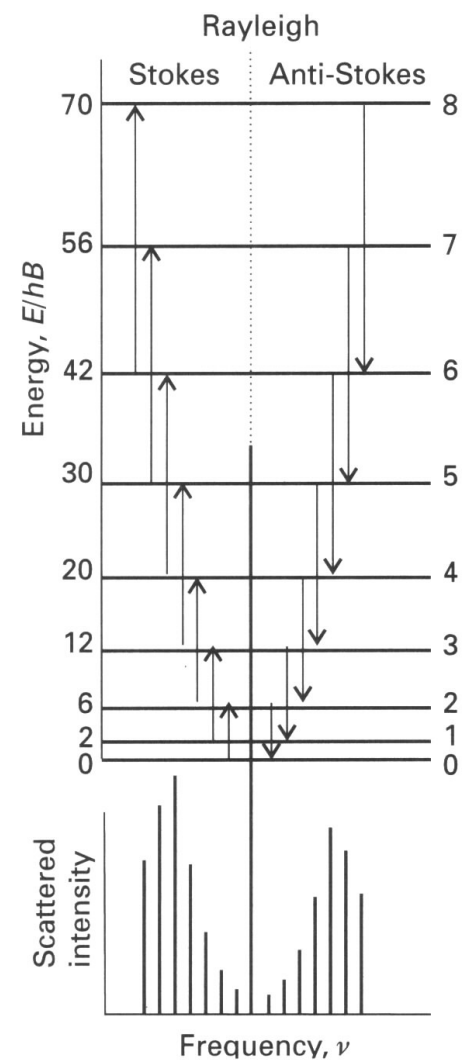


Rotational Raman spectroscopy

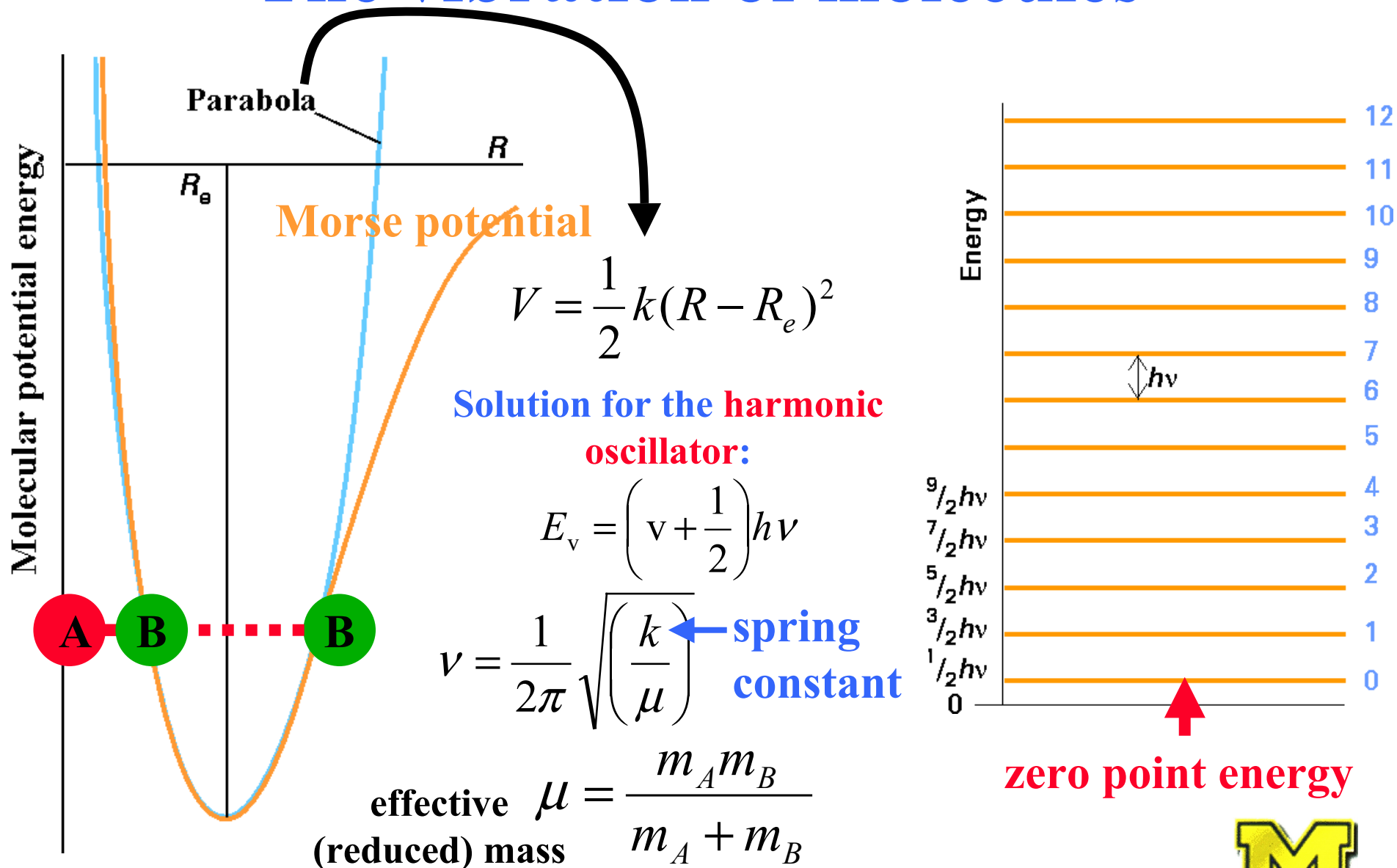


Gross selection rule:
anisotropic polarization
(example: H-H!)

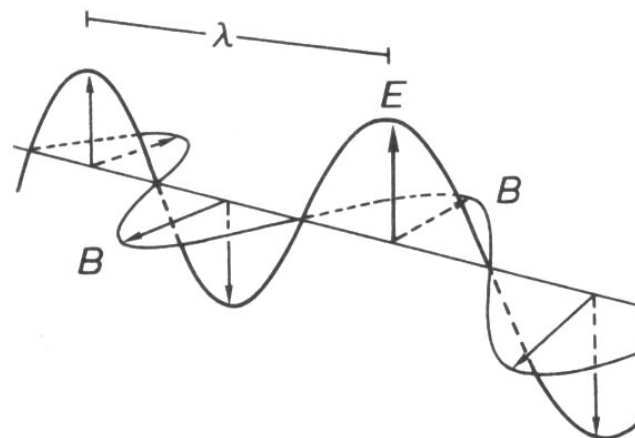
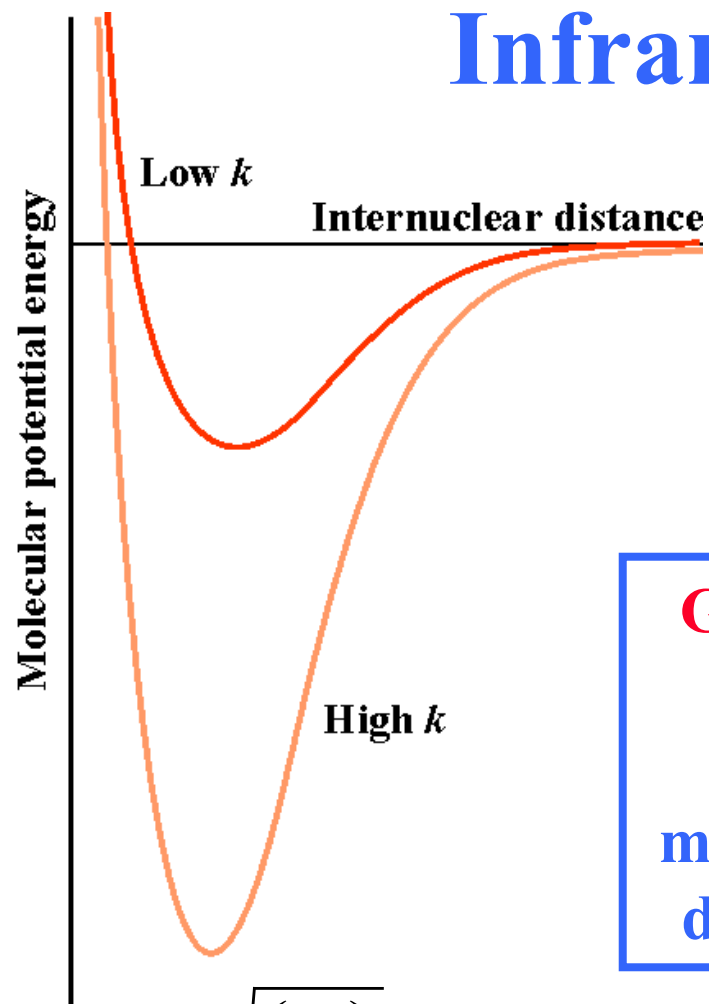
Specific selection rules:
 $\Delta J = +2$ (Stokes lines)
 $\Delta J = -2$ (anti-Stokes lines)



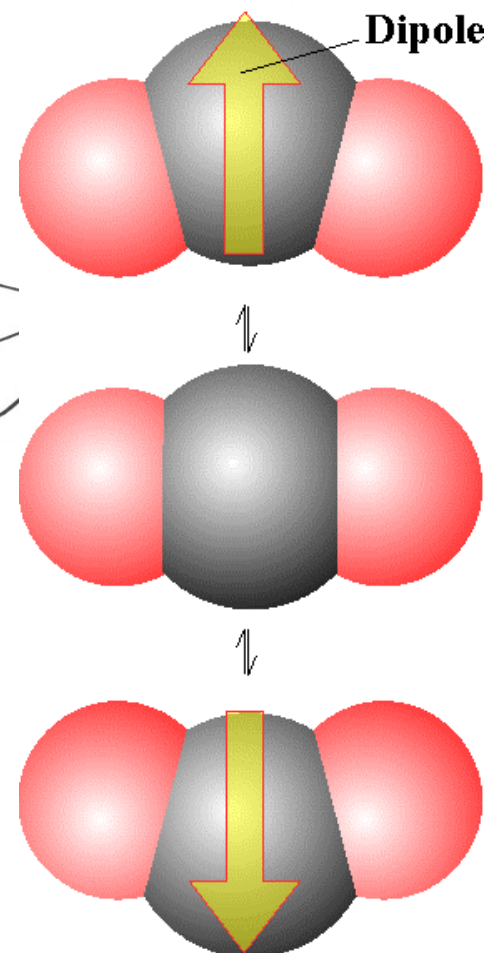
The vibration of molecules



Vibrational transitions: Infrared spectroscopy



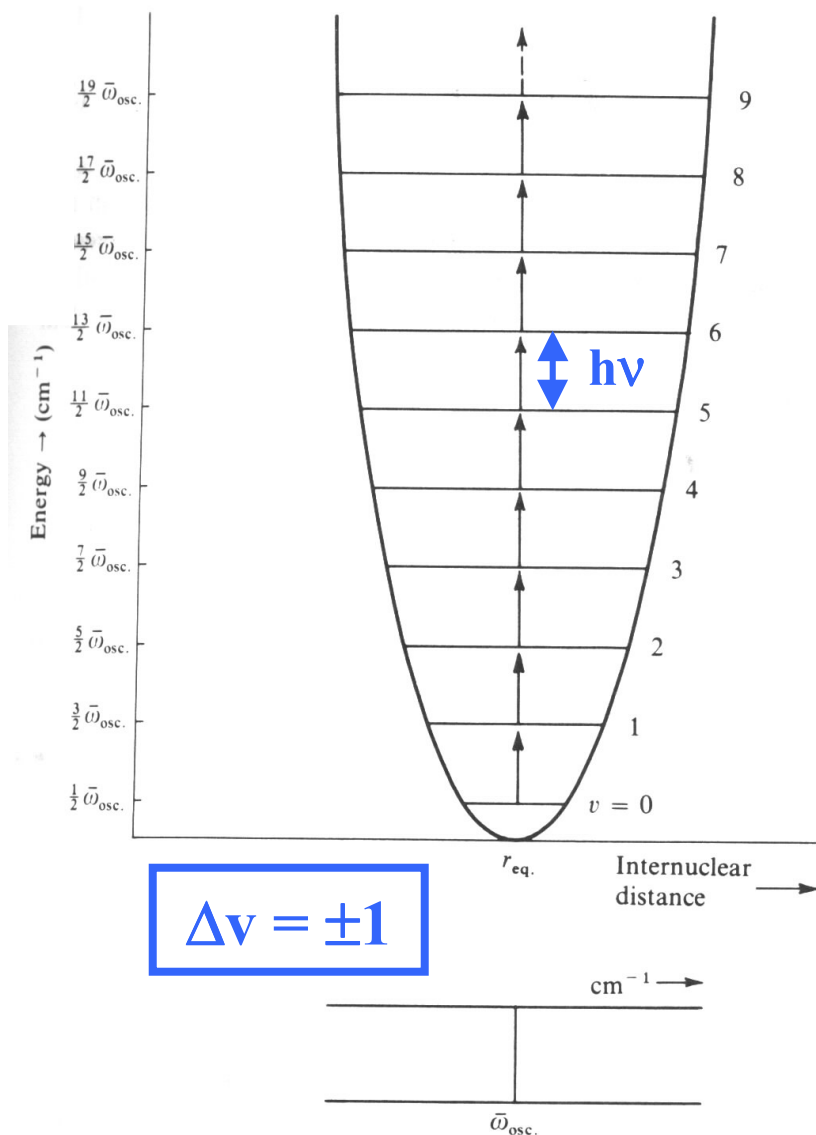
Gross selection rule:
the electric dipole
moment of the
molecule must change
during the vibration



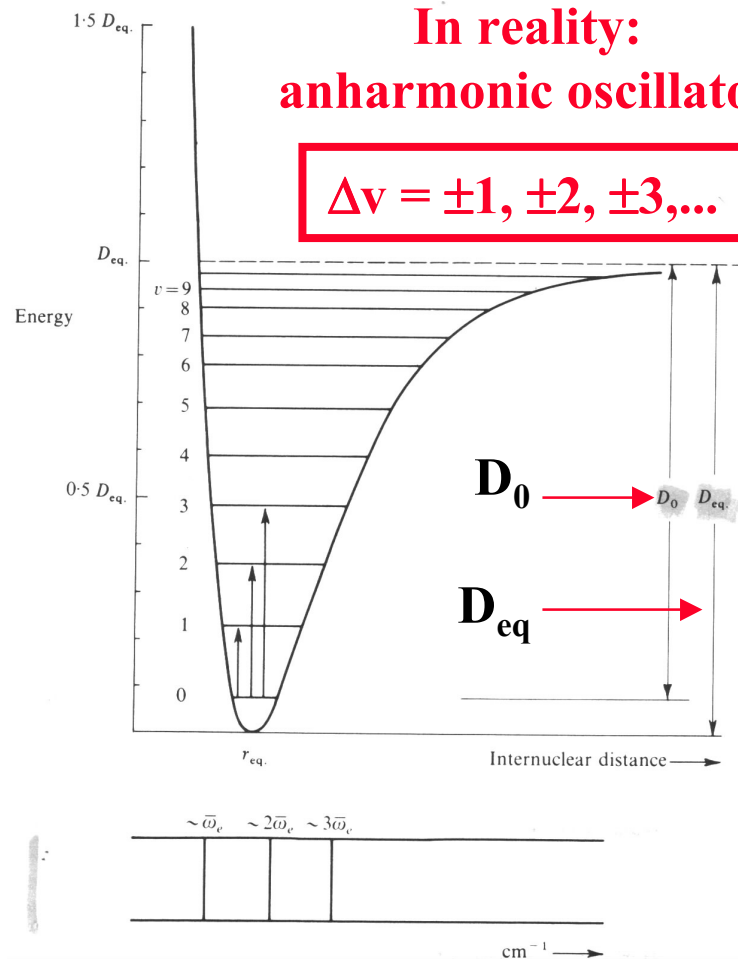
$$\nu = \frac{1}{2\pi} \sqrt{\left(\frac{k}{\mu}\right)} \Rightarrow 300-3000 \text{ cm}^{-1} = \text{Infrared}$$

Specific selection rules

harmonic oscillator



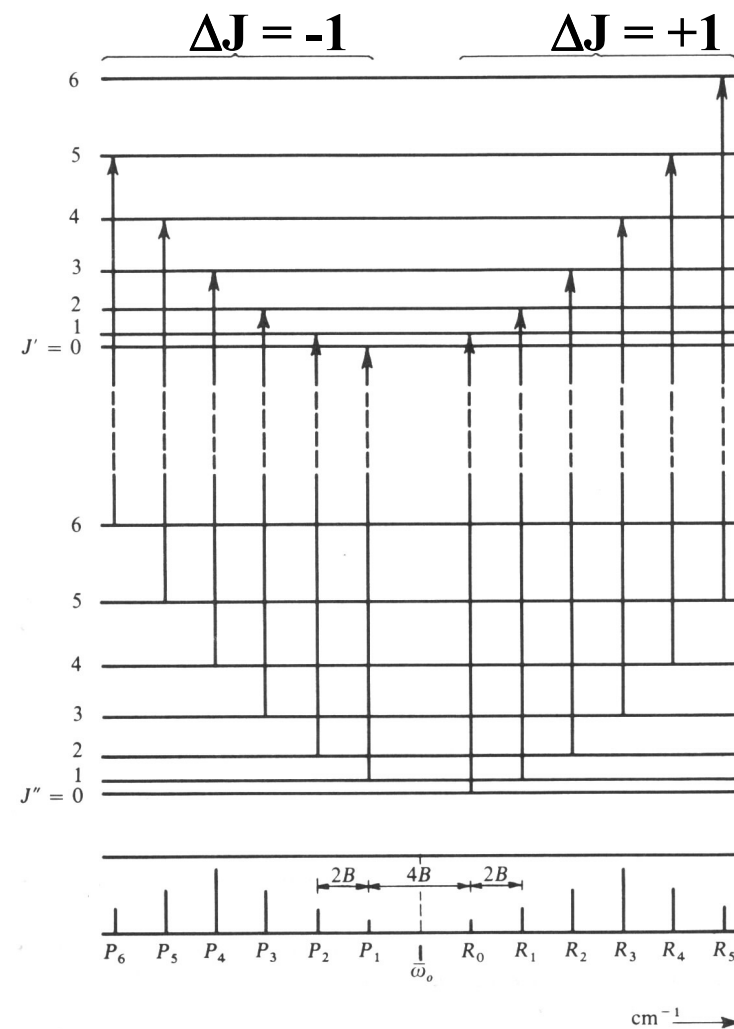
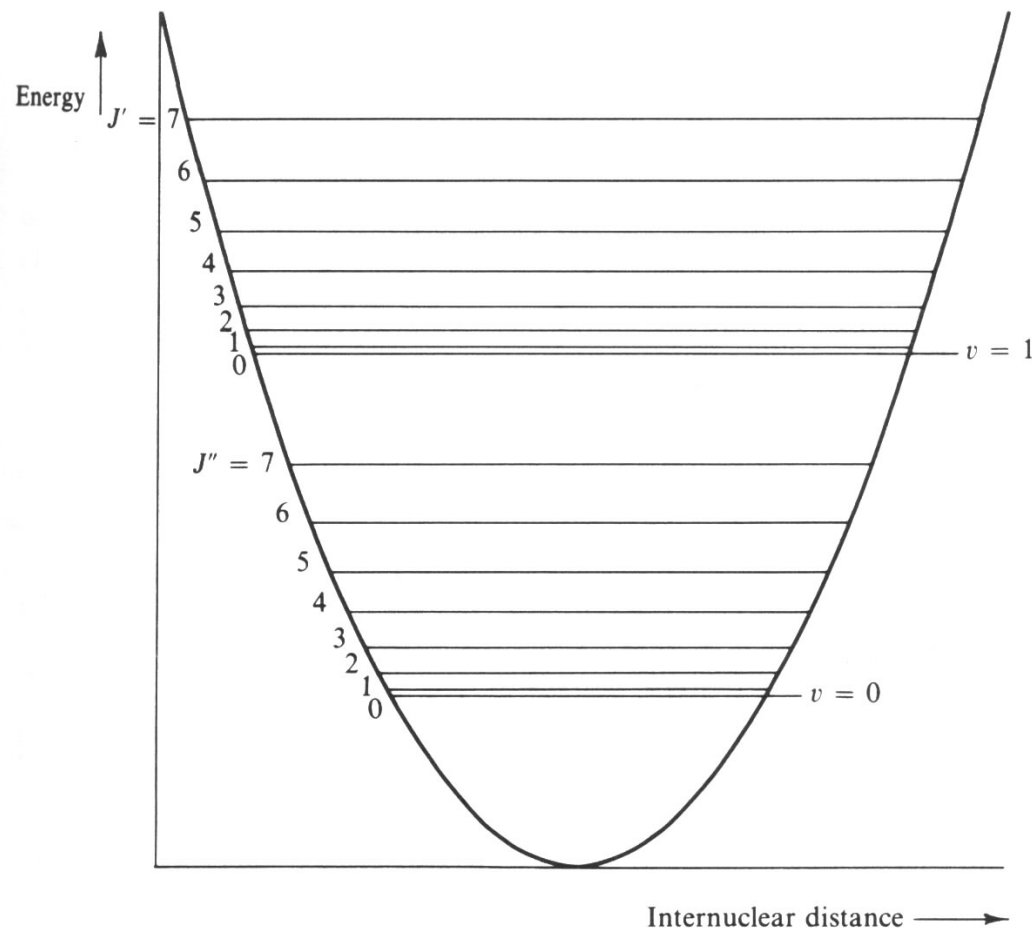
In reality: anharmonic oscillator



x_e = anharmonicity constant

$$E_v = \left(v + \frac{1}{2} \right) h\nu - \left(v + \frac{1}{2} \right)^2 h\nu x_e$$

The vibrating rotor



Born-Oppenheimer approximation:

The energies of rotations and vibrations are so different that $E_{\text{total}} = E_{\text{rot.}} + E_{\text{vib.}}$



Vibrations of polyatomic molecules: How many are there?

Each atom can move along one of three axes:
 $\Rightarrow 3N$ possible displacements (= degrees of freedom)



Three of these degrees of freedom correspond to translational motion:
 $\Rightarrow 3N - 3$ degrees of freedom left



Three (/two) degrees of freedom correspond to rotations:
 $\Rightarrow 3N - 6$ ($3N - 5$ for linear molecule) degrees of freedom left for vibrations

