

Relating Structure and Function in Two Homologous Cytochromes c

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Electron transfer is a fundamental reaction in biological energy transduction. Redox potential, \mathcal{E}° , is a major factor determining electron transfer energetics and rates. In this project, we aim to understand the structural basis for the different redox potentials of two homologous bacterial cytochromes c, *Hydrogenobacter thermophilus* cytochrome *c*₅₅₂ (*Ht* cyt *c*₅₅₂) and *Pseudomonas aeruginosa* cytochrome *c*₅₅₁ (*Pa* cyt *c*₅₅₁).

The redox potential of *Ht* cyt *c*₅₅₂ (220 mV vs. NHE) is low relative to other proteins in the cyt *c*₈ subfamily, and is 65 mV lower than that of *Pa* cyt *c*₅₅₁. Primary sequences (69% similarity) and tertiary structures (0.7-Å RMSD) of *Pa* cyt *c*₅₅₁ and *Ht* cyt *c*₅₅₂ are highly homologous. Surprisingly, the major structural determinants of redox potential (axial ligands to the heme iron, percent solvent exposure of the heme group, interactions with heme propionates, and electrostatic potential at the heme iron active site) predict a higher redox potential for *Ht* cyt *c*₅₅₂ relative to *Pa* cyt *c*₅₅₁. Hydrogen exchange (HX) studies, however, reveal that despite conservation of the fold of the Cys-X-X-Cys-His motif, there are substantial differences in the energetics of local unfolding events for residues within and near the motif between *Pa* cyt *c*₅₅₁ and *Ht* cyt *c*₅₅₂. Analysis of the disparate HX behavior suggests that local unfolding events are more energetically accessible in *Ht* cyt *c*₅₅₂ than in *Pa* cyt *c*₅₅₁. From this analysis, we hypothesize that the energetically accessible local unfolding events that occur in *Ht* cyt *c*₅₅₂ allow for transient solvent access to the heme active site, which could explain the lower redox potential of *Ht* cyt *c*₅₅₂ compared to *Pa* cyt *c*₅₅₁. We thus propose that energetics of protein structural fluctuations near redox cofactors may play a role in modulating metalloprotein redox potential.