

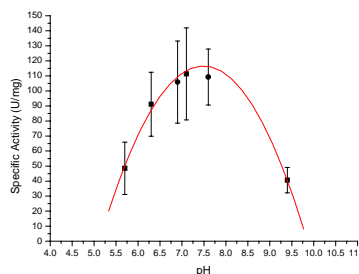
Mechanism of Nitrous Oxide Reduction with Activated N₂O Reductase from *Achromobacter Cycloclastes*

Koyu Fujita, David M. Dooley

Department of Chemistry and Biochemistry, Montana State University

Denitrification is the reductive cascade from nitrate (NO₃⁻), nitrite (NO₂⁻), nitric oxide (NO), nitrous oxide (N₂O) to dinitrogen (N₂) and therefore is an intrinsic process of the global nitrogen cycle. N₂O is kinetically inert but is thought to contribute to ozone depletion and global warming. Therefore, it is important to achieve a comprehensive understanding of N₂O metabolism. N₂O reductase (N₂OR) catalyzes the two electron reduction of N₂O to N₂ and H₂O. This is the terminal step of the denitrifying process. According to the recent crystallographic results on homodimeric N₂ORs from *Pseudomonas nautica*¹ and *Paracoccus denitrificans*², this enzyme contains two copper centers, Cu_A and Cu_Z. The Cu_A site in N₂OR is quite similar to that in cytochrome c oxidase and is implicated in electron transfer to the active site. On the other hand, the Cu_Z site is a unique structural motif consisting of μ₄-sulfide bridged tetranuclear Cu cluster. Although the Cu_Z site is thought to function as a catalytic center to reduce N₂O, the mechanism of N₂O reduction is not well understood. Recently, we isolated a recombinant N₂OR from *Achromobacter cycloclastes* (Ac) and found it is significantly activated towards N₂O reduction by reduced methyl viologen (MV). Moreover, the activated Ac N₂OR is less colored and has a low intensity EPR signal after preincubation with reduced MV.³ These results indicate that the catalytically active state of Cu_Z in Ac N₂OR is 4[Cu(I)].

Herein, pH dependence of the activation and the reactivity of the activated Ac N₂OR towards the reaction with N₂O will be shown in order to shed light on the mechanism of N₂O reduction in the N₂OR. Preliminary results of the pH profile suggest the activity of the Ac N₂OR is enhanced under neutral pH conditions. The plausible mechanism of N₂O reduction with N₂OR will be discussed on the basis of the data. In addition, influence of anion species such as chloride and azide towards the activity of N₂ORs will be taken into consideration.



1. Brown, K.; Tegoni, M.; Prudencio, M.; Pereira, A. S.; Moura, J. J.; Moura, I.; Cambillau, C. *Nat. Struct. Biol.* **2000**, 7, 191.
2. Haltia, T.; Brown, K.; Tegoni, M.; Cambillau, C.; Saraste, M.; Mattila, K.; Djinovic-Carugo, K. *Biochem. J.* **2003**, 369, 77.
3. Chen, J. M.; Bollinger, J. A.; Dooley, D. M. *J. Am. Chem. Soc.* **2004**, 126, 3030.