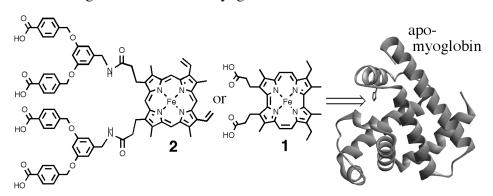
Unusual Dioxygen Affinity of Myoglobin Reconstituted with an Artificially Created Iron Complex

<u>Takashi Hayashi^{1,2}</u>, Takashi Matsuo^{1,2}, Hideaki Sato², Dai Murata^{1,2}, and Yoshio Hisaeda²

¹Department of Applied Chemistry, Graduate School of Engineering, Osaka University, Suita, Osaka 565-0871, Japan, and ²Department of Chemistry and Biochemistry, Faculty of Engineering, Kyushu University, Fukuoka 812-8581, Japan.

Functionalization of hemoproteins is one of the attractive subjects for creating a new biomaterial. Recently, we have prepared various artificial prosthetic groups and inserted them into apomyoglobin to obtain reconstituted myoglobins. For example, a porphycene iron complex 1 as shown below is a unique prosthetic group. Compared to native myoglobin, the reconstituted myoglobin with 1 is clearly stable against acid denaturation, indicating the Fe–His93 bond strength is strong. The O_2 affinity of the ferrous 1 is higher by 2600-fold than that of the wild-type myoglobin, mainly due to the decrease in the O_2 dissociation rate. In addition, the M' value, which is a ratio of CO/O_2 affinity, is less than 1. On the other hand, a reconstituted myoglobin having 2 with a hydrophobic domain at the terminal of the two heme-propionate side chains shows unusual CO/O_2 discrimination. The substantial ligand selectivity for the reconstituted myoglobin remarkably increases in favor of O_2 over CO with the M' value of 0.88, indicating that the modification of heme-propionate side chains only perturbed CO affinity. These data suggest that the incorporation of an artificially created heme will lead to an effective regulation of a ligand binding. In this presentation, we summarize our recent results of ligand binding studies using the reconstituted myoglobins.



- 1) T. Matsuo, T, Hayashi *et al. J. Am. Chem. Soc.* **2004**, *126*, 16007– 16017.
- 2) H. Sato, T. Hayashi *et al. J. Am. Chem. Soc.* **2005**, *127*, 56–57.