Study of covalent and non-covalent interactions in [Pd(dien)nucleobase]²⁺/I-tryptophan(N-Acetyl-tryptophan) systems: Formation of metal-tryptophan species by nucleobase substitution under biologically relevant condition.

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Substitution of the model nucleobases: 1-Methylcytosine (1-MeCyt) and 9-Ethylguanine (9-EtGH) by the amino acid l-tryptophan (Trp) and N-acetyltryptophan (N-AcTrp) was observed in complexes $[Pd(dien)(1-MeCyt)]^{2+}$ (2) and $[Pd(dien)(9-EtGH)]^{2+}$ (3) respectively, under physiological pH. The resulting species $[Pd(dien)(Trp)]^{2+}$ (6) and $[Pd(dien)(N-AcTrp)]^{2+}$ (7) were characterized using ^{1}H -, ^{13}C - NMR and ESI-MS techniques as coordinated through the amino and deprotonated amido nitrogen respectively. Complexes (6) and (7) were also obtained from a solution of [Pd(dien)Cl]Cl (1) and Trp or N-AcTrp respectively, to confirm the proposed structure. Substitution reactions did not occur in analogs Pt(II) complexes namely $[Pt(dien)(1-MeCyt)]^{2+}$ (4) and $[Pt(dien)(9-EtGH)]^{2+}$ (5) suggesting that the intrinsic more reactive nature of the Pd(II) center is the responsible for the substitution phenomenon. The association constant of complexes (2-5) was further investigated towards N-AcTrp under conditions that ruled out covalent bonding to examine both: molecular recognition through π - π stacking of the nucleobases with the indole ring in N-AcTrp, and the effect of the nature of the metal. The solid state structure for (2) and (4) was also determined and compared with analogous Pt, Pt-cytosine complexes.

Scheme I. Substitution reactions for Pd-nucleobase complexes with 1-tryptophan.