

Synthesis of Iron/Sulfur Clusters in Non-Polar Solvents

- Toward Better Clusters Modeling the Nitrogenase Active Sites -

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The iron sulfide proteins in nature are represented by those consisting of [4Fe-4S], [3Fe-4S], and [2Fe-2S] clusters. In most of the cases, these clusters are stable, and the core geometries are readily synthesized by spontaneous self-assembly reactions in polar aprotic solvents. We have recently found a new methodology to construct Fe sulfide clusters, which utilizes the bis-amide complex of iron, $\text{Fe}\{\text{N}(\text{SiMe}_3)_2\}_2$, as the precursor. This new route has allowed us to build iron/sulfur clusters in non-polar solvents, and resulted in a series of unprecedented cluster structures, some of which resemble closely the active sites of nitrogenases.

The unusual [8Fe-7S] inorganic core structure of the reduced form of P-cluster (P^{N}) was self-assembled by the reaction of $\text{Fe}\{\text{N}(\text{SiMe}_3)_2\}_2$, tetramethylthiourea (*tmtu*), 2, 4, 6-triisopropylbenzenethiol (HS-*tip*), and elemental sulfur in the ratio of 8(Fe):3(*tmtu*):12(HS-*tip*):7(S). Interestingly the complex $[\{\text{N}(\text{SiMe}_3)_2\}\{\text{SC}(\text{NMe}_2)_2\}\text{Fe}_4\text{S}_3]_2(\mu_6\text{-S})\{\mu\text{-N}(\text{SiMe}_3)_2\}_2$ (**1**), obtained therefrom, consists of irons with a 6Fe(II)+2Fe(III) oxidation state as manifested by the Mössbauer study, while P^{N} has been thought to carry 8 Fe(II) ions. Although cluster **1** degrades to a Fe_4S_4 cubane cluster under the presence of thiols, as the P-cluster does so, it was possible to synthesize $(\text{NEt}_4)_2[\{\text{N}(\text{SiMe}_3)_2\}(\text{SAr})\text{Fe}_4\text{S}_3]_2(\mu_6\text{-S})\{\mu\text{-N}(\text{SiMe}_3)_2\}_2$ (**2**) (Ar = 4-^tBu- C_6H_4 , 2-SiMe₃- C_6H_4 etc.) from the reactions of **1** with corresponding ammonium salts of thiolates. We have also isolated $[(\text{S-dmp})\text{Fe}_4\text{S}_3]_2(\mu_6\text{-S})(\mu\text{-S-dmp})_2\{\mu\text{-N}(\text{SiMe}_3)_2\}$ (**3**) from the reaction of $\text{Fe}\{\text{N}(\text{SiMe}_3)_2\}_2$, HS-dmp, HS-*tip*, and elemental sulfur (8:6:10:7), which is topologically analogous to FeMo-co and has an intriguing 5Fe(II)+3Fe(III) oxidation state.

